

Geological Carbon Sequestration Research Publications

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CO₂ and Co-injectant Solubility (impurities in CO₂ streams)

- Ji X and **Zhu C** (2013) Predicting possible effects of H₂S impurity on CO₂ transportation and geological storage. *Environmental Science & Technology* 47: 55-62, dx.doi.org/10.1021/es301292n.
- Ji X and **Zhu C** (2012) A SAFT Equation of State for the Quaternary H₂S-CO₂-H₂O-NaCl system. *Geochimica et Cosmochimica Acta* 91: 40–59, doi.org/10.1016/j.gca.2012.05.023.
- Ji, X. and **Zhu, C.** (2010) Modelling of phase equilibria in the H₂S-H₂O system with statistical associating fluid theory. *Energy & Fuels* v. 24, 6208-6213. DOI:10.1021/ef100847j.
- Hu, J, Duan, Z, **Zhu, C.**, and Chou, I., (2007) PVTx properties of the CO₂-H₂O and CO₂-H₂O-NaCl systems below 647K: Assessment of experimental data and thermodynamic models. *Chemical Geology*, v. 238, p.249-267.
- Duan, Z., Sun, R, Liu, R, and **Zhu, C.** (2007) Accurate thermodynamic model for H₂S solubility in pure water and brines. *Energy & Fuels*, v. 21, 2056-2065.
- Duan, Z., Sun, R., **Zhu, C.** and Chou, I., (2006) An improved model for the calculation of CO₂ solubility in aqueous solutions containing Na⁺, K⁺, Ca²⁺, Mg²⁺, Cl⁻, and SO₄²⁻. *Marine Chemistry*. v.98, 131-139.

Geochemical Modeling of CO₂ Reactions with Reservoir and Cap Rocks

- Zhang GR, *Lu P, Huang Y, Li GH, ***Zhu C** (2021) Investigation of mineral trapping processes based on coherent front propagation theory: A dawsonite-rich natural CO₂ reservoir as an example. *International Journal of Greenhouse Gas Control* v110, 103400, <https://doi.org/10.1016/j.ijggc.2021.103400>.
- ‡Zhang GR, ‡Lu P, ‡Zhang YL, Wei XM, ***Zhu C** (2016) Impacts of mineral reaction kinetics and regional groundwater flow on long-term CO₂ fate at Sleipner. *Energy & Fuel* 30(5): 4159-4180, doi: 10.1021/acs.energyfuels.5b02556s
- **Zhu C**, ‡Zhang GR, ‡Lu P, ‡Meng LF, Ji X (2015) Benchmark modeling of the Sleipner CO₂ plume: Calibration to seismic data for the uppermost layer and model sensitivity analysis. *The International Journal of Greenhouse Gas Control* 43: 233-246, doi: 10.1016/j.ijggc.2014.12.016
- Zhang GR, ‡Peng L, ‡Zhang YL, Wei XM, ***Zhu C** (2015) Effects of rate law formulation on predicting CO₂ sequestration in sandstone formations. *International Journal of Energy Research* 39(14): 1890-1908, doi: 10.1002/er.3374.
- Liu, Faye (Yifei), P. Lu, C. **Zhu**, Y. Xiao (2011) Coupled reactive transport modeling of CO₂ Sequestration in the Mt. Simon Sandstone Formation, Midwest U.S.A. *The International Journal of Greenhouse Gas Control*, DOI: 10.1016/j.ijggc.2010.08.008.
- Strazisar, B. R., **Zhu, C.**, and Hedges, S. W., (2006) Preliminary modeling of the long-term fate of CO₂ following injection into deep geological formations. *Environmental Geosciences* v.13, no.1, 1-15, 2006.

Mineralogical or above Ground Carbon Sequestration (U.S. Patent 7922792 "Method for Sequestration of CO₂ and SO₂ Utilizing a Plurality of Waste Streams")

- Dilmore, R., Lu, P., Soong, Y. , Allen, Hedges, H., Fu, J. K., Dobbs, C., Degalbo, A., **Zhu, C.** , (2008) Sequestration of CO₂ in mixtures of bauxite residue and saline waste water. *Energy & Fuels*, v 22, No.1, p.343-353.

Laboratory Experiments on Host Rock Reactivity and Caprock Integrity

- Lu P, Fu Q, Seyfried Jr. WE, Jones K., and Zhu C (2013) Coupled alkali feldspar dissolution and secondary mineral precipitation in batch systems: 2 Effects of CO₂ and implications for carbon sequestration. *Applied Geochemistry* 30: 75-90, [Doi10.1016/j.apgeochem.2012.04.005](https://doi.org/10.1016/j.apgeochem.2012.04.005)
- Liu, Faye (Yifei), P. Lu, C. Griffith, S.a W. Hedges, C. **Zhu** (2012) CO₂-caprock-brine interaction: Reactivity Experiments on Eau Claire Shale and a review of literature. *The International Journal of Greenhouse Gas Control* 7: 153–167, doi.org/10.1016/j.ijggc.2012.01.012.
- Lu, P, Fu, Q., Seyfried, W.E. Jr, Hereford, A.G., **Zhu**, C. (2010) Navajo Sandstone-Brine-CO₂ interaction: Implications for Geological Carbon Sequestration. *Environmental Earth Sciences*, 62 (1): 101-118doi:10.1007/s12665-010-0501-y

Single Mineral Reaction Kinetics with Applications to CCUS (last five years)

- ‡Kang JT, Bracco JN, Rimstidt JR, Zhu GH, Huang F, ***Zhu** C (2022). Ba attachment and detachment fluxes to and from barite surfaces in ¹³⁷Ba-enriched solutions with variable [Ba²⁺]/[SO₄²⁻] ratios near solubility equilibrium. *Geochimica et Cosmochimica Acta*. v317, 180-200, doi.org/10.1016/j.gca.2021.11.008.
- **Zhu** C, ‡Zhang YL, Rimstidt JD, Gong L, ‡Burkhart, JA, Chen KY, Yuan HL (2021) Testing hypotheses of albite dissolution mechanisms at near-equilibrium using Si isotope tracers. *Geochimica et Cosmochimica Acta*. v303, 15-37. doi.org/10.1016/j.gca.2021.03.023.
- **Zhu** C, Rimstidt JD, ‡Zhang YL, ‡Kang JT, Schott J, Yuan HL (2020) Decoupling feldspar dissolution and precipitation rates at near-equilibrium with Si isotope tracers: Implications for modeling silicate weathering. *Geochimica et Cosmochimica Acta*. v271, 132-153. DOI: 10.1016/j.gca.2019.12.024.
- Gong L, Rimstidt JD[†], ‡Zhang YL, Chen KY, ***Zhu** C (2019) Unidirectional kaolinite dissolution rates at near-equilibrium and near-neutral pH conditions. *Applied Clay Science* v182, doi.org/10.1016/j.clay.2019.105284;
- Zhang YL, Gong L, Chen KY, ‡Burkhart J, Yuan HL, ***Zhu** C (2020) A method for Si isotope tracer kinetics experiments: Using Q-ICP-MS to obtain ²⁹Si/²⁸Si ratios in aqueous solutions. *Chemical Geology* v531, Doi: 10.1016/j.chemgeo.2019.119337;
- Rimstidt, JD, ‡Zhang Y, ***Zhu** C (2016) Rate equations for sodium catalyzed amorphous silica dissolution. *Geochimica et Cosmochimica Acta* 195: 120-125. doi.org/10.1016/j.gca.2015.07.030
- **Zhu** C, ‡Liu ZY, ‡Wang C, ‡Schaefer A, ‡Lu P, ‡Zhang GR, ‡Zhang YL, Georg RB, Rimstidt JD, Yuan HL (2016) Measuring silicate mineral dissolution rates using Si isotope doping. *Chemical Geology*, 445: 146-163, [doi:10.1016/j.chemgeo.2016.02.027](https://doi.org/10.1016/j.chemgeo.2016.02.027)

Geochemical Modeling Tools

- Lu P, Zhang GR, Apps J, ***Zhu** C. (2022) Comparison of thermodynamic data files for PHREEQC. *Earth-Science Reviews*, <https://doi.org/10.1016/j.earscirev.2021.103888>..
- Zhang GR, Lu P, ‡Zhang YL, ‡Tu K, ***Zhu** C (2020) SupPHREEQC: A program to generate customized PHREEQC thermodynamic database based on Supcrtbl. *Computers & Geosciences*. v143. doi.org/10.1016/j.cageo.2020.104560.
- ‡Zhang YL, ‡Hu B, Teng YG, ***Zhu** C (2019) A library of BASIC scripts of reaction rates for geochemical modeling using PHREEQC. *Computers & Geosciences*, v133, doi.org/10.1016/j.cageo.2019.104316;
- ‡Zimmer K, ‡Zhang YL, ‡Lu P, ‡Chen YY, ‡Zhang GR, ***Zhu** C (2016) SUPCRTBL: A revised and extended thermodynamic dataset and software package of SUPCRT92. *Computers & Geosciences* 90: 97-111, [doi:10.1016/j.cageo.2016.02.013](https://doi.org/10.1016/j.cageo.2016.02.013).

Online Geochemical Modeling Tools (3000 visits between 9/20 and 11/21)

- CO₂ Solubility Calculator: Users input temperature, pressure, and NaCl molality, and the online tool generates CO₂ solubility.

- H₂S Solubility Calculator: H₂S is a common impurity in CO₂ streams in CCUS. Users input temperature, pressure, NaCl molality, and H₂S and CO₂ mole fractions, the online tool generates H₂S solubility.
- Online PHREEQC: We developed an online version of the US Geological Survey modeling software for speciation – solubility reaction path, and one-dimensional coupled reactive transport modeling.
- Online SUPCRTBL: An online version of SUPCRT that is dynamically linked to a MySQL database to make preparation of input files easier.